Roll No. 22001

M. Sc. (Chemistry) 2nd Semester Examination – May, 2019

INORGANIC CHEMISTRY-II

Paper : CY(H)-201

Time: Three hours !

[Maximum Marks : 80

Before answering the questions, candidates should ensure that they have been supplied the correct and complete question paper. No complaint in this regard, will be entertained after examination.

Note: Question No. 1 is compulsory and all questions carry equal marks. Attempt five questions in all, selecting at least one question from each Section.

1. Compulsory Question:

2+2+2+2+2+2+2=16

- (a) What homonuclear and heteronuclear carbonyls?
- What is Nephelauxetic effect?
- Explain Synergic effect.

P. T. O.

https://www.haryanapapers.com

https://www.haryanapapers.com

22001

- (d) What is Neil's temperature?
- (e) What is G.S.T. for Mn²⁺ ion?
- (f) Predict structure of [Ru₅N(CO)₄] using skeletal and non-skeletal electrons.
- (g) Differentiate between atomic orbital and molecular orbital.
- (h) Define T.I.P.

SECTION - A

https://www.haryanapapers.com

8

- 2. (a) Draw and explain M.O. diagram for the octahedral complex [Cr(H₂O)₆]³⁺. 8
 - (b) Discuss all the factors influencing the stability of coordination compounds. 8
- 3. (a) Discuss Crystal Field Theory.
 - (b) Discuss effect of π -overlapping on the magnitude splitting energy, considering only overlapping with dxy orbitals of metal used. 8

(2)

https://www.haryanapapers.com

SECTION - B

- (a) Discuss electronic spectra for molecular (compounds of l₂.
 - (b) Discuss Orgel diagrams for d² and d⁴ meters for octahedral complexes. Also predict pretransitions.
- (a) Discuss and draw the T-S diagram for d⁵ sym
 - (b) Explain LMCT transitions in tetrahedral comp taking suitable examples.
 - (c) Explain John Tellor distortion with sulf examples.

SECTION - C

- 6. (a) Explain Guoy's method for determination magnetic susceptibility.
 - (b) What are L.N.C.C ? Also discuss structure bonding in following : (μ-CO)₂ (CpRh)₃CO and η⁴-(C₄H₄)₂ Fe₂(CO)₃

(3)

P. T. **Q**,

https://www.haryanapapers.com

https://www.haryanapapers.com

7. (a) What is magnetic moment? Also discuss orbital contribution to magnetic moment of substances for high spin and low spin complexes.

001

(h) Draw and discuss structure of $B_{10}H_{14}$ and meta- $C_2B_2H_{12}$.

SECTION - D

- (n) Explain bonding in carbonyl complexes. 6
- (b) Discuss the structure of sodium nitroprusside. 5
- (c) Write short note on tertiary phosphine as ligand. 5
- (a) Describe linear and bent Nitrosyl Ligand with suitable examples.
- (b) Why dinitrogen is weaker ligand in comparison to carbonyl ligands? Explain by M.O. diagrams of each.
 8

(4)