

Roll No.

24005

**B. Tech. 2nd Semester
(Common for all Branches) (Re-Appeal)
Examination – October, 2020**

ENGINEERING CHEMISTRY

Paper : CH-101F

Time : 1.45 hours]

| Maximum Marks : 100

Before answering the questions, candidates should ensure that they have been supplied the correct and complete question paper. No complaint in this regard, will be entertained after examination.

Note : Attempt any **three** questions. All questions carry equal marks.

1. (a) Define the system having incongruent melting.
- (b) Define metastable equilibrium.
- (c) Differentiate triple point and eutectic point.
- (d) Define Break-point chlorination.
- (e) Define demineralization of water.
- (f) Describe stress cracking.
- (g) What do you understand by galvanization ?
- (h) Describe saponification value of a lubricant.
- (i) Write uses of teflon.
- (j) What do you understand by Blue-shift ?

24005-1950-(P-3)(Q-9)(20)

P. T. O.

2. (a) How is the phase diagram of water helpful in explaining (i) Ice skating (ii) Flow of glaciers.

b (b) Explain the mechanism of Homogeneous and heterogeneous catalysis.

3. (a) Draw and explain the phase diagram of sodium sulphate- water system.

b (b) Explain the concepts of promoters, inhibitors and poisoners.

4. (a) 100 ml of water sample requires 10 ml N/50 EDTA when titrated using buffer solution of pH 9-10 and EBT indicator, calculate the hardness of water. https://www.mdustudy.com

(b) What do you understand by desalination of water ? Discuss in detail the reverse- osmosis process for desalination of sea water with help of neat, clean and labeled diagram. What are its advantages and disadvantages ?

5. (a) Write short notes on :

- (i) Boiler corrosion
- (ii) Caustic embrittlement.

(b) A zeolite softener was 90% exhausted by removing the hardness completely when the 200000 litres of hard water sample passed through it. The exhausted zeolite bed requires 145 litres of 35% NaCl solution for its complete regeneration. Calculate the hardness of water.

24005-1950-(P-3)(Q-9)(20)

(2)

6. (a) Write short notes on :
- (i) Waterline corrosion
 - (ii) Role of sacrificial anode in corrosion control.
- (b) Write short notes on :
- (i) Molybdenum disulphide as solid lubricant.
 - (ii) Semi-solid lubricants.
7. (a) Why additives are used in lubricants ? Gives some examples of additives/which are commonly used in lubricants.
- (b) Write short notes on :
- (i) Dry corrosion
 - (ii) Microbial corrosion
8. (a) Discuss the principle and application of TGA
- (b) Write short notes on :
- (i) Polymer composite
 - (ii) Buna-S
9. (a) Write the applications of U.V. and I.R. spectroscopy. 16
- (b) Write short notes on : 8
- (i) Ziegler-Natta Catalyst
 - (ii) Write uses of Teflon and Phenol-formaldehyde resin